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1. State the number of particles in a mole and how it relates to Avogadro's number.

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moles, & particles. Molarity p. 412-415. Explain the meaning of molarity and how it is used as a conversion factor. CHEMISTRY TEST STUDY GUIDE

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What is the mass of 8.40 moles of magnesium

Nov 20, 2010 What is the mass of 8.40 moles of magnesium chloride, Questions about chemistry study guide?

I-iv l2 wb fm (se)-824344

The mass of a mole of hydrogen atoms is 1.00×10^{23} amu. Study Guide for Content Mastery Chemistry: Matter and Change

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